

PETRYUCHENKO, A.Ya.

Best section in the Debal'tsevo Division. Avtom. telem. i svyaz' 3
no.8:23-24 Ag '59. (MIRA 13:2)

1. Starshiy elektromekhanik Debal'tsevskoy distantzii signalizatsii i
svyazi Donetskoy dorogi.

(Railroads--Signaling)

(Railroads--Block system)

PETRYSZYN, C.; SZARKOWSKA, L.

Trehalase activity during metamorphosis of the hawk moth *Celerio euphorbiae* L. *Bul Ac Pol biol* 7 no.12:491-493 '59. (EEAI 9:12)

1. Institute of Biochemistry and Biophysics, Polish Academy of Sciences. Presented by J.Heller.

(HAWK MOTHS) (CELERIO EUPHORBIAE) (TREHALASE)

MOCHNACKA, Irena; PETRYSZYN, Czesława

Trehalose in *Celerio euphorbiae*. Acta biochim. polon. 6 no.3:
307-311 '59.

1. Zakład Biochemii Ewolucyjnej Instytutu Biochemii i Biofizyki
Pan. Zakład Chemii Fizjologicznej A.M. Warszawa.
(DISACCHARIDES chem.)
(INSECTS)

PETRYSZYN, Jerzy; LEWCZUK, Zofia

Studies on the durability of the preparation of pepsin. Przem spos
16 no.4:68-72 Ap '62

1. Centralne Laboratorium Przemyslu Paszowego, Pracownia Biochemiczna.

S

CONTRIBUTION TO THE PROBLEM OF PREPARATION OF MOULDING SANDS. L. Petrálek. (Hunticke Listy, 1950, vol 5, Sept., pp 363-369; C. t., pp. 406-412). In Czech. The author reviews the deficiencies of natural sands and the objects of sand preparation and reports the results of tests on processes of preparing sand. The influence of drying on the properties of green glauconite sand was determined after drying for 4 hr. at the temperatures 90°, 110°, 300° and 400° C. samples which had been prepared for periods of 5, 10, and 20 min. in a Simpson mixer. Compression strength and permeability decreased with increasing drying temperatures, but the decrease was reduced by a longer mixing time; the moisture content corresponding to optimum compression strength remained practically unchanged. The compression strength of specimens mixed in a pit type mixer, changed gradually from 900g/sq. cm. to up to 1250 g/sq. cm. after 1 hr. mixing; this reduced the optimum moisture content from about 6% to 4% after a mixing time of 60 min. In the case of sands mixed with bentonite maximum compression strength was obtained after mixing for 10 to 20 min.---E.G.

ASB-15A METALLURGICAL LITERATURE CLASSIFICATION

15000 15100 15200 15300 15400 15500 15600 15700 15800 15900 16000 16100 16200 16300 16400 16500 16600 16700 16800 16900 17000 17100 17200 17300 17400 17500 17600 17700 17800 17900 18000 18100 18200 18300 18400 18500 18600 18700 18800 18900 19000 19100 19200 19300 19400 19500 19600 19700 19800 19900 20000 20100 20200 20300 20400 20500 20600 20700 20800 20900 21000 21100 21200 21300 21400 21500 21600 21700 21800 21900 22000 22100 22200 22300 22400 22500 22600 22700 22800 22900 23000 23100 23200 23300 23400 23500 23600 23700 23800 23900 24000 24100 24200 24300 24400 24500 24600 24700 24800 24900 25000 25100 25200 25300 25400 25500 25600 25700 25800 25900 26000 26100 26200 26300 26400 26500 26600 26700 26800 26900 27000 27100 27200 27300 27400 27500 27600 27700 27800 27900 28000 28100 28200 28300 28400 28500 28600 28700 28800 28900 29000 29100 29200 29300 29400 29500 29600 29700 29800 29900 30000 30100 30200 30300 30400 30500 30600 30700 30800 30900 31000 31100 31200 31300 31400 31500 31600 31700 31800 31900 32000 32100 32200 32300 32400 32500 32600 32700 32800 32900 33000 33100 33200 33300 33400 33500 33600 33700 33800 33900 34000 34100 34200 34300 34400 34500 34600 34700 34800 34900 35000 35100 35200 35300 35400 35500 35600 35700 35800 35900 36000 36100 36200 36300 36400 36500 36600 36700 36800 36900 37000 37100 37200 37300 37400 37500 37600 37700 37800 37900 38000 38100 38200 38300 38400 38500 38600 38700 38800 38900 39000 39100 39200 39300 39400 39500 39600 39700 39800 39900 40000 40100 40200 40300 40400 40500 40600 40700 40800 40900 41000 41100 41200 41300 41400 41500 41600 41700 41800 41900 42000 42100 42200 42300 42400 42500 42600 42700 42800 42900 43000 43100 43200 43300 43400 43500 43600 43700 43800 43900 44000 44100 44200 44300 44400 44500 44600 44700 44800 44900 45000 45100 45200 45300 45400 45500 45600 45700 45800 45900 46000 46100 46200 46300 46400 46500 46600 46700 46800 46900 47000 47100 47200 47300 47400 47500 47600 47700 47800 47900 48000 48100 48200 48300 48400 48500 48600 48700 48800 48900 49000 49100 49200 49300 49400 49500 49600 49700 49800 49900 50000 50100 50200 50300 50400 50500 50600 50700 50800 50900 51000 51100 51200 51300 51400 51500 51600 51700 51800 51900 52000 52100 52200 52300 52400 52500 52600 52700 52800 52900 53000 53100 53200 53300 53400 53500 53600 53700 53800 53900 54000 54100 54200 54300 54400 54500 54600 54700 54800 54900 55000 55100 55200 55300 55400 55500 55600 55700 55800 55900 56000 56100 56200 56300 56400 56500 56600 56700 56800 56900 57000 57100 57200 57300 57400 57500 57600 57700 57800 57900 58000 58100 58200 58300 58400 58500 58600 58700 58800 58900 59000 59100 59200 59300 59400 59500 59600 59700 59800 59900 60000 60100 60200 60300 60400 60500 60600 60700 60800 60900 61000 61100 61200 61300 61400 61500 61600 61700 61800 61900 62000 62100 62200 62300 62400 62500 62600 62700 62800 62900 63000 63100 63200 63300 63400 63500 63600 63700 63800 63900 64000 64100 64200 64300 64400 64500 64600 64700 64800 64900 65000 65100 65200 65300 65400 65500 65600 65700 65800 65900 66000 66100 66200 66300 66400 66500 66600 66700 66800 66900 67000 67100 67200 67300 67400 67500 67600 67700 67800 67900 68000 68100 68200 68300 68400 68500 68600 68700 68800 68900 69000 69100 69200 69300 69400 69500 69600 69700 69800 69900 70000 70100 70200 70300 70400 70500 70600 70700 70800 70900 71000 71100 71200 71300 71400 71500 71600 71700 71800 71900 72000 72100 72200 72300 72400 72500 72600 72700 72800 72900 73000 73100 73200 73300 73400 73500 73600 73700 73800 73900 74000 74100 74200 74300 74400 74500 74600 74700 74800 74900 75000 75100 75200 75300 75400 75500 75600 75700 75800 75900 76000 76100 76200 76300 76400 76500 76600 76700 76800 76900 77000 77100 77200 77300 77400 77500 7

S/081/62/000/020/030/040
B160/B144

AUTHOR: Petrykowski, Andrzej

TITLE: Properties of polyamides and injection molding parameters

PERIODICAL: Referativnyy zhurnal. Khimiya, no. 20, 1962, 502, abstract
20P74, (Mechanik, 1961, v. 34, no. 10, 517 - 519 [Pol.])

TEXT: The technological parameters of injection molding relate to temperature of the polyamide, temperature of the mold, cooling time, pressure, injection rate. The effect of these and of the wall thicknesses of articles on the shrinkage of polyamide articles, i.e. on the accuracy of their dimensions, is discussed. Studies of six types of polyamide produced in Poland and elsewhere are given. [Abstracter's note: Complete translation.]

Card 1/1

PETRYL, J.

GEOGRAPHY & GEOLOGY

PERIODICAL: CESKY LID. Vol. 42, no. 5, 1955

PETRYL, J. E. Nohejlova-Pratova's Krasa ceske mince (The Beauty of Czech Coins);
a book review, p. 235.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, No. 2, Feb 50, Unclass.

PETRÝL, J.

GEOGRAPHY & GEOLOGY

PERIODICAL: ČESKOSLOVENSKÁ ETHNOGRAFIE. Vol. 6, no. 4, 1958.

PETRÝL, J. Commercial importance of the hawkers' prints in the Chrudim printing houses during the 19th century. p. 396.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, No. 2,
February 1959, Unclass.

PETRYS, Tibor, dr. adjunkt

The Rebinder effect. Wiad chem 16 no.7:433-448 J1 62.

1. Zaklad Chemii Technicznej, Politechnika, Krakow.

AUTHORS: Petrys, Tibor and Datka, Maria POL/5-58-23-4/30

TITLE: Graphite and Disulphide of Molybdenite Application to Anti-Friction Lubricants (Zastosownie grafitu i dwusiarczku molibdenu do smarów przeciwtarciowych)

PERIODICAL: Przegląd Techniczny, 1958, Nr 23, pp 1099-1101 (POL)

ABSTRACT: The authors discuss the application of graphite and molybdenite in anti-friction lubricants. They describe friction encountered by metals (Table I) lubricated with various anti-friction lubricants. This article further gives "N.P. Petrov" friction formula and the research results of the molybdenite lubricants discussed at the International Mechanics Engineers Convention held in Bad-Ragaz, Switzerland. The Polska Katedra Fizyki Politechnika Kraków (Faculty of Physics, Polytechnic in Kraków) conducts research on molybdenite lubricants. ✓

Card 1/1

PETRYA, T.

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 // Chemistry of the lubrication of metals. Tibor Petrya,
Chemik (Gliwice) 12, 104-8 (1959).—The following groups of
 lubricants are discussed: those which do not react with the
 surface of the metal (hydrocarbons); activated lubricants
 contg. additives, such as org. acids, their salts, esters, alde-
 hydes, and CO and S contg. compds., lubricants with fillers,
 such as graphite and MoS_2 . P. J. Hendel
 5/11/81

NETREK, I. [Netrek, I.; G. YA, V.; ...]

Relation between the latent period of the reflex potential in
 agyris suprasylvialis nucleus and gross motor parameters after
 and the intensity of peripheral excitation. Izv. eksp. b.
 1 mod. 55 / 1.0. 55 / no. 12-10 ...

1. Iz laboratorii vysshey nerovny deyatel'nosti. - korr.
 korrespondent AN - chekhalovatskoy ...
 prof. Ya. Grbek) i kafedry anatomii ...
 1. Trzavy) meditsinskogo fakulteta Universiteta imeni ...
 Olomouts, chekhalovatskaya Sotsialisticheskaya Respublika.

Peckel A, L.

Contribution to the Study of the Adhesiveness of Moulding
Materials to Castings. L. Petrálek. (Reports of Czech
Foundry Research: Supplement, 1957, 2, 2), Supplemental.
[In Czech]. Castings in the shape of split cylinders were
not round specimens of moulding materials. It was found
that baking-on of sand to gray cast iron was minimized by
adding anthracite dust. Graphite and coke dust proved
ineffective. Coal dust additions are not suitable for steel
castings.—p. 2.

of

PETRZELA, L.

Dry-molding mixtures with quartz sands. Prace. p. 89.
SLEVARENSTVI, Prague, Vol. 2, no. 12, Dec. 1954.

SO: Monthly List of East European Accessions, (EEAL), LC, Vol. 5, No. 6,
June 1956, Uncl.

PETRZELA, LEV.

Slevarenske formovaci latky. (Vyd. 1.) Praha, Statni nakl. technicke literatury, 1955. 63p. (Molding materials for foundries. 1st ed. illus., bibl., indexes, tables)

SOURCES: East European Accessions List (EEAL), LC, Vol 5, no. 3, March 1956

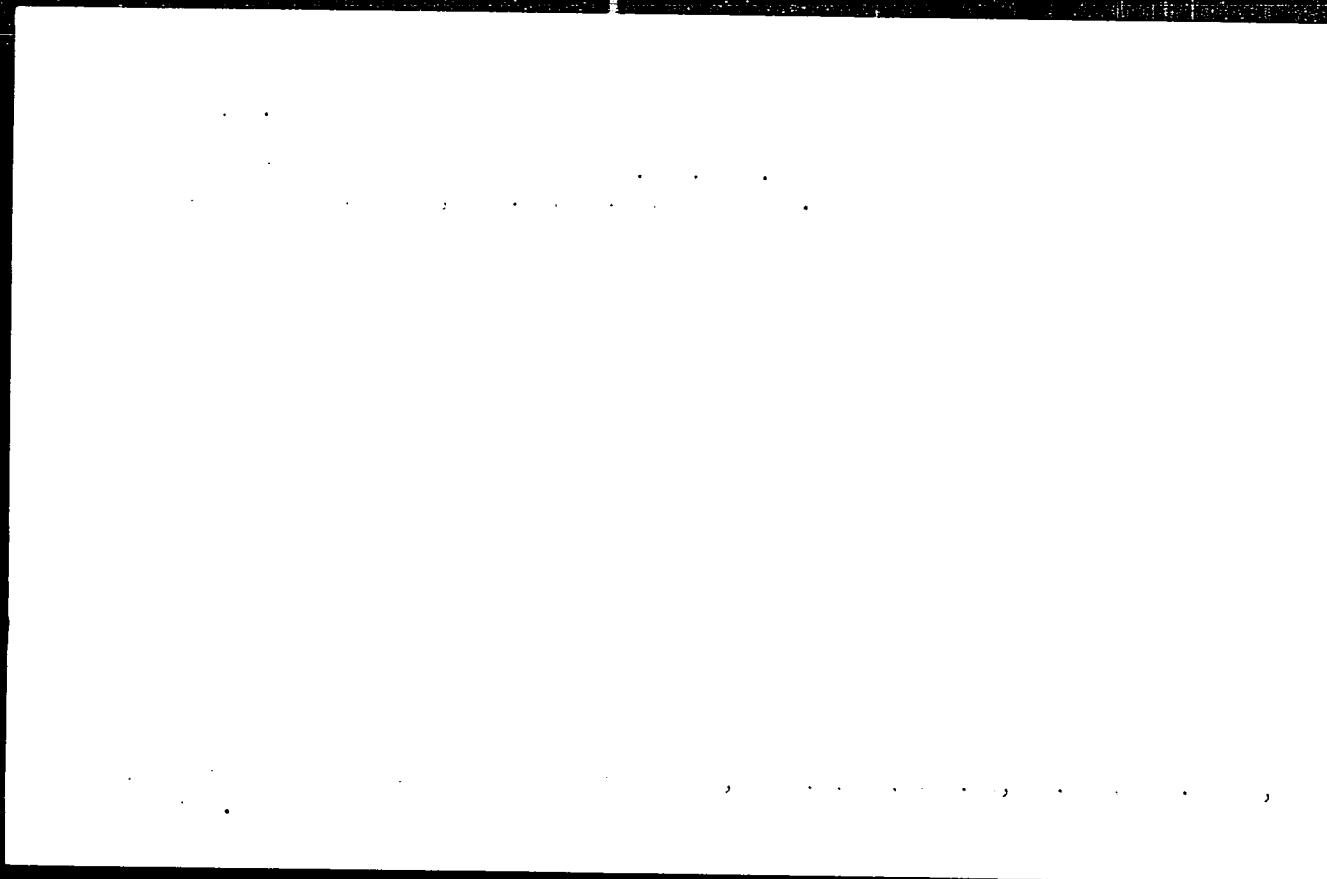
PETRZELA, L.

Discuss on of J. Stejskal's article "Actual State of Exploitation of Rajec Sands." p. 67, SLEVARŮNSTVÍ (Ministerstvo strojírenství a Ministerstvo hutního průmyslu a rudných dolů) Praha, Vol. 3, No. 3, Mar. 1955

SOURCE: East European Accessions List (EEAL) Library of Congress, Vol. 4, No. 12, December 1955

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[illegible]

1. *Journal of the American Medical Association*, 2000; 283: 2689-2693.

PETRZELA, L.

"Exothermic sleeves for risers. Prace.p.39"

SLEVARENSTVI. (Ministerstvo tezkého stromdrenstvi a Cezkoslovenska vedecka
technicka spolenost pro hutnictvi a slevarenstvi) Praha, Czechoslovakia,
Vol. 3, No. 8 Aug. 1955.

Monthly List of East European Accessions (EEAI), LC, Vol. 8, No. 6 June 1959
Uncl.

PETRZELA, L.

____ Petrzela, L. Casting Produced by shell molding. p. 163.

SO: Monthly List of the East European Scissions. (EEAL). LC. Vol. 4,
no. 10, Oct. 1955. Uncl.

PETRZELA, L.

Petrzela, L. Casting Produced by shell molding. p.196

SO; Monthly List of the East European Accessions. (REAL). LC. Vol. 4,
no. 10, Oct. 1955. Uncl.

PETRZELA, L.

Effect of molding materials on solidification rate of castings. p. 121.

SLEVARENSTVI Vol. 4, no. 4, Apr. 1956

Czechoslovakia

Source: EAST EUROPEAN LISTS Vol. 5, no. 7 July 1956

PETRZELA, L.

Notes
✓ Properties of mould materials at elevated temperatures. L. Petržela
(Slodrenskul, 1958, 4, Rep. Czech. Foundry Res., 285-284).
Highest compressive strength at elevated temp. with slow heating
was obtained with clay binders at 900-1000°. The thermal expansion
of such mixtures is determined by the sharp sand in the mixture.
In chemically hardened mixtures, max. compressive strength occurs
between 450 and 550°, owing to combination of hydrated SiO₂ and
Na₂CO₃. Natrification of Ca-bentonites increased strength at elevated
temp. by as much as three times but had no effect on thermal
expansion. (From English summary). J. S. C.

ITM... ..

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Eleven
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PETRŽELA, LEV.

✓ The production of chemically hardened molds (CO₂ method). Lev Petržela. *Freiberg. Forsch. I-RT, 218-27 (1956)*.—This article briefly summarizes the present knowledge of the chemically setting mixts. which consist of sand and water glass, and which are set by CO₂. The principle of chem. setting is the development of hydrated SiO₂ by decomg. the aq. alkali silicate soln. The chem. reaction between water glass and CO₂ is given as an example: $R_2O + nSiO_2 + mH_2O + CO_2 \rightarrow R_2CO_3 \cdot pH_2O + n(SiO_2 \cdot pH_2O)$. The sand mixts. which are used for molds and cores are given for steel castings, for gray and malleable cast iron, and for nonferrous and light metal castings. The precautions required in treating the raw material and the details of prepn. and storing of the mixts. are mentioned. A detailed account is given of the effect of admixts. which significantly improve and those which harm the chemically setting mixts. The effect of the kind of sand used are discussed, such as particle size, grain shape, clay cement content, mineralogical purity of sand particles above 0.1 mm., and dust content (below 0.06 mm.). The refractory properties of these chemically set mixts. are given as well as their strength after heating and cooling. Thermal expansion is considered. The importance of regenerating all the waste material is stressed, and the details of recovery are briefly described. These chemically set sands represent an important supplement to the casting industry, and their use is increasingly expanded. 63 references.

George Meister

1971, 1.

Next to radical - One time with the ... 1971

1. (Stow master) in ... , June 1, 1971, Collection 1

SO: Monthly Index of East Europe - Accession ... 1. ... 11 November 1971

MRZELA, L.

Effect of heat on brains of man.

J. 2/4 (Slevarenstii) Vol. 1, no. 4, Sept. 1968, Praha, Czechoslovakia

SO: MONTHLY INDEX OF EAST EUROPEAN ACQUISITIONS (MEAI) 10, VOL. 9, NO. 3, JAN. 1971

PETRZELA, I.

"Shell molding. Prace."

p. 99 (Slevarenstvi) Vol. 5, no. 12, Dec. 1957.
Prague, Czechoslovakia

SO: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4,
April 1958

PETRZELA, Lev

New device for determining permeability. Slevarenstvi 11 no.8/9:
316-319 Ag '63.

1. Statny vyzkumny ustav materialu a technologie, slevarensky
vyzkum, Brno.

PETRZELKA, K.

Prospects for mechanizing and automatizing work in the textile industry. p. 123.
(Textil, Praha, Vol. 9, no. 4, Apr. 1954)

SO: Monthly list of East European Accessions (SEAL), LC Vol 4, No. 6, June 1955, Uncl

PETRZELKA, K.

Losses during the introduction of new types of products. p. 339. (Textil, Praha,
Vol. 9, no. 11, Nov. 1954)

SO: Monthly list of East European Accessions (EEAL), LC Vol 4, No. 6, June 1955, Uncl

PETRZELA L.

Chemically set foundry sands. L. Petrzela (Vysokom
 octavy, tsikho, strolimstvi, Brno, 1934). P. prep. foundry mixes
 for rapid setting by the addn. of water glass (I) to sand,
 e.g., sand 93, coal dust 5, water 1, and 36-38° B_é 1 4.5
 5.0 kg. are mixed for 5-10 min. Total water content should
 be 3.4-4.0%. Av. compn. of I was Na₂O 7.4-7.6, SiO₂
 25.0-26.6, CaO 1.4-1.6, MgO 0.4, and other metallic oxides 0.3%.
 The setting is due to the action of CO₂ as per equation:
 Na₂SiO₃.aq. + CO₂ → Na₂CO₃.aq. + SiO₂.aq. The mixes
 after setting have compression strength 8 kg./sq. cm.
 By the use of these mixes foundry output can be increased
 2-10 times. Frank J. Hendel

PETRZELA, L.

~~USSR~~

13084* Castings Produced by Shell Molding. Výroba odlitků do křehkoplastových forem. (Czech.) Lev Petráň. Strojárství, v. 3, no. 7, July 1955, p. 198-202. ~~-----~~
Review of production methods, including molding techniques and testing. Photograph, diagrams, graphs.

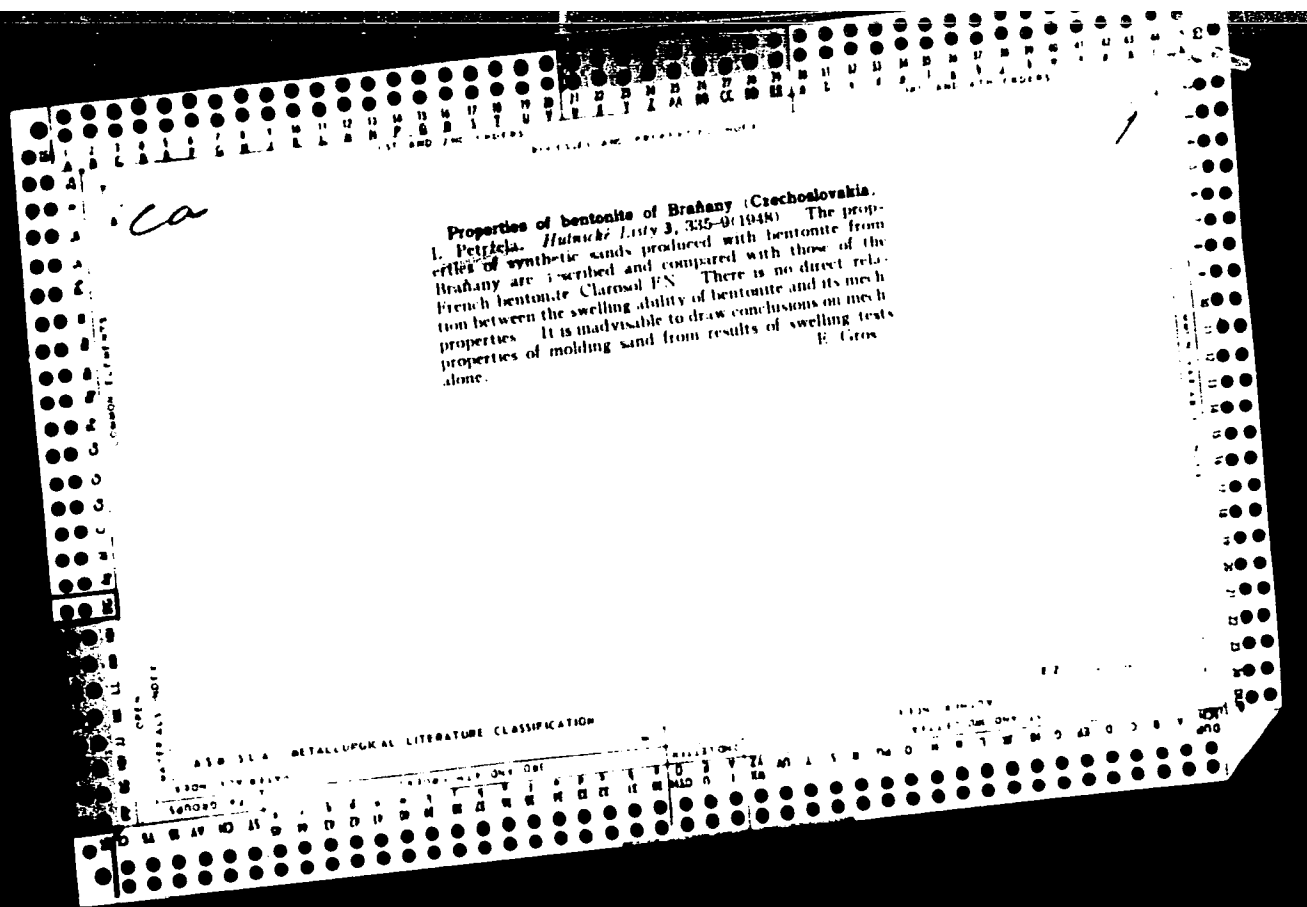
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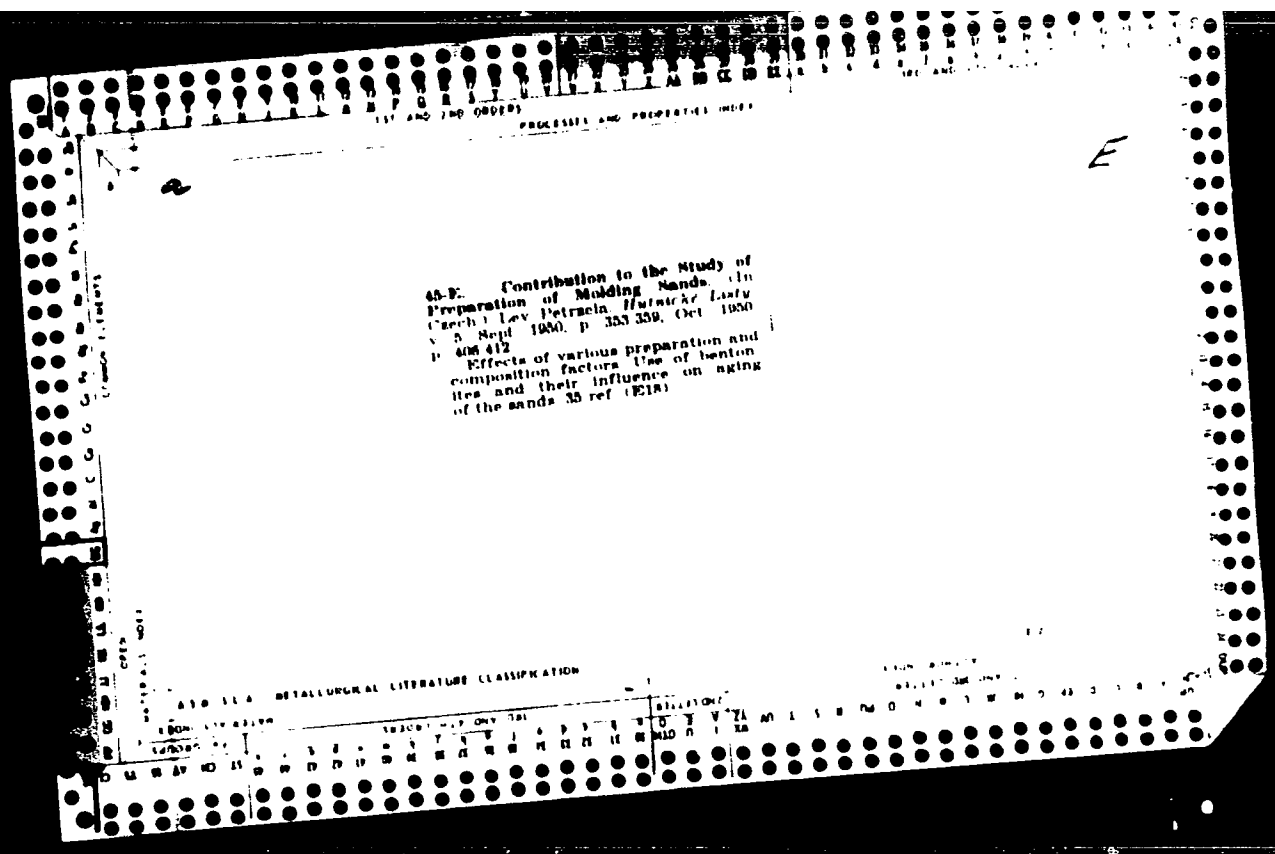
PETRAZELA, Lev

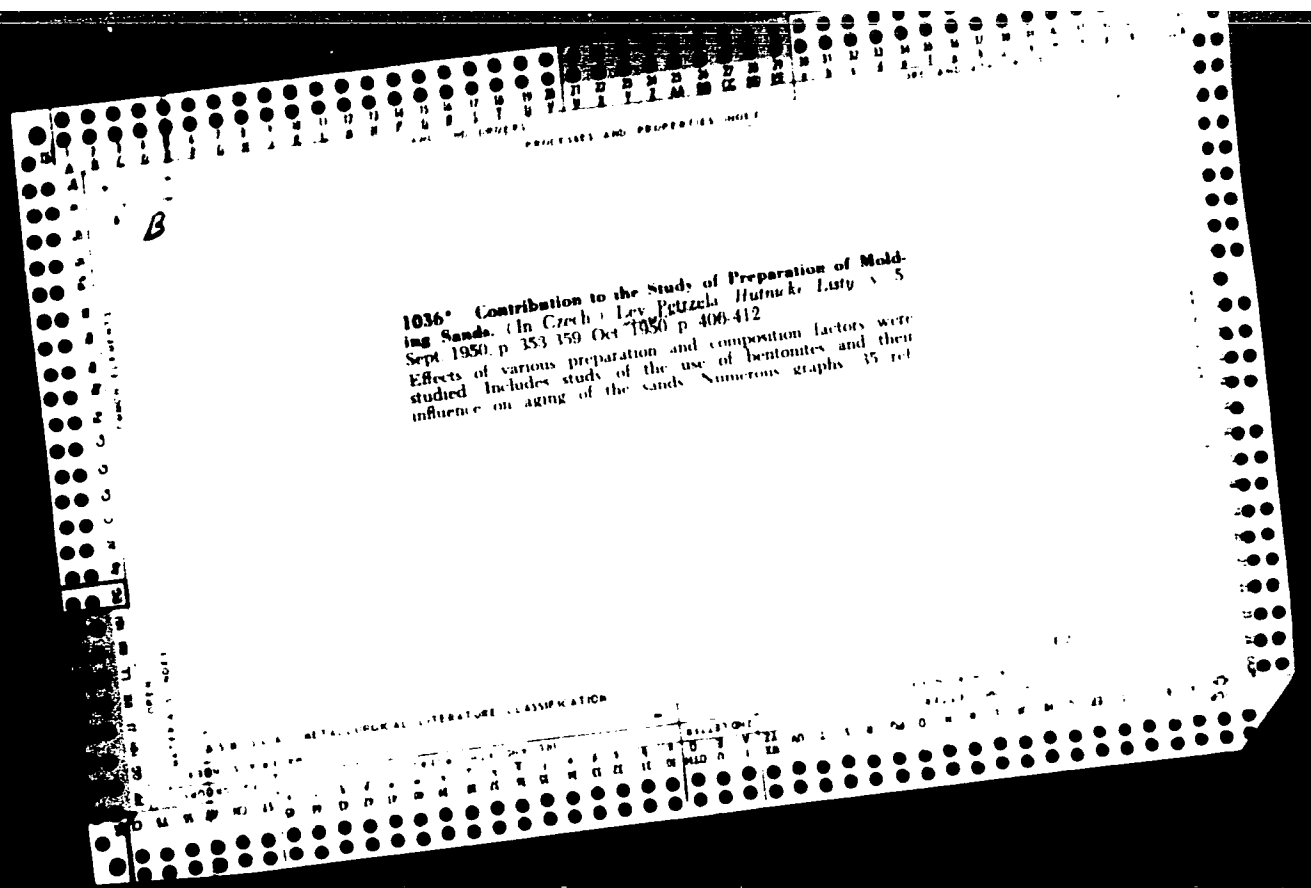
Prosperity of science and industry research. Slevarenstvi 13
1944-1945

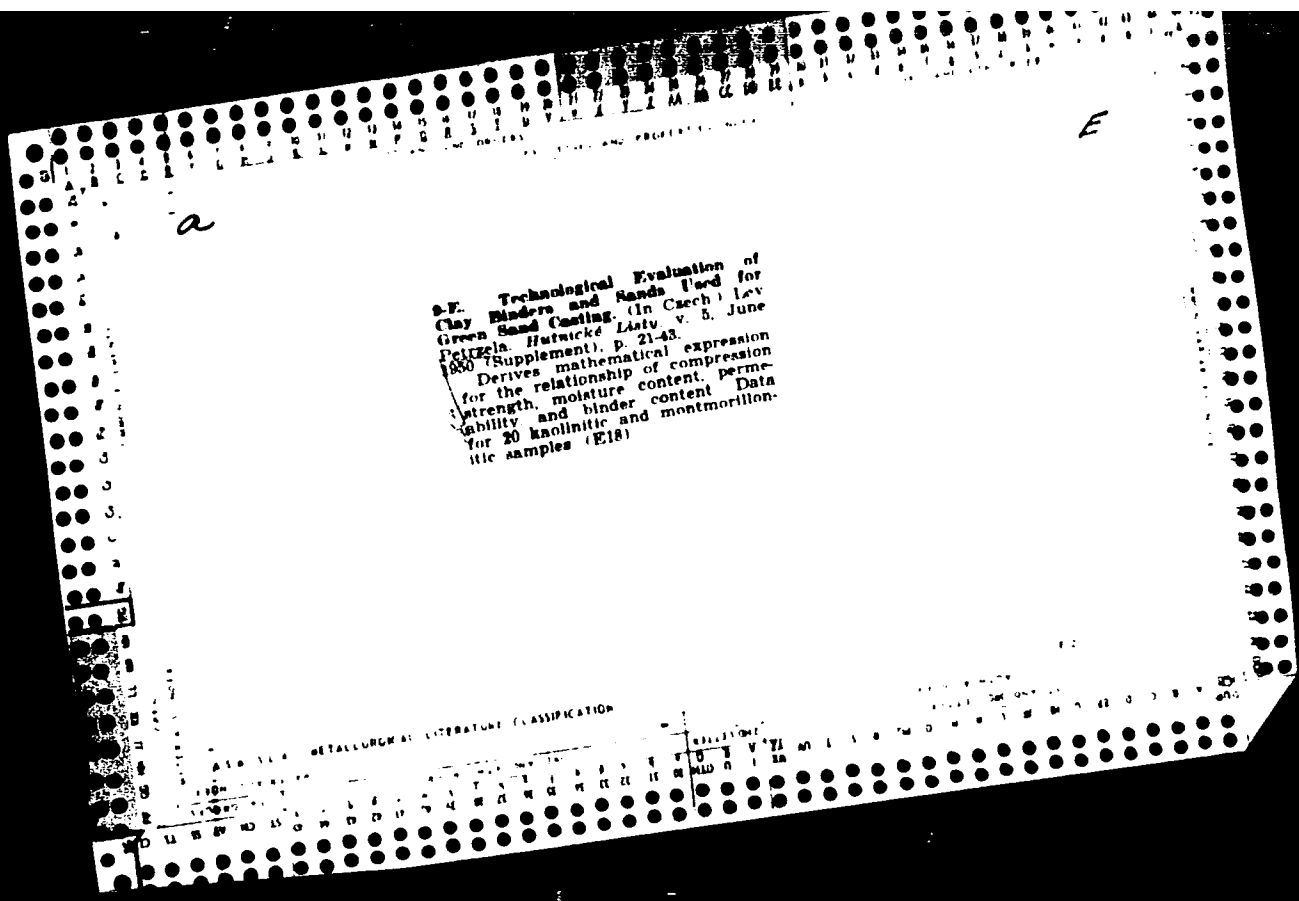
1. State Scientific Institute of Material and Technology, Foundry
Research 1944-1945

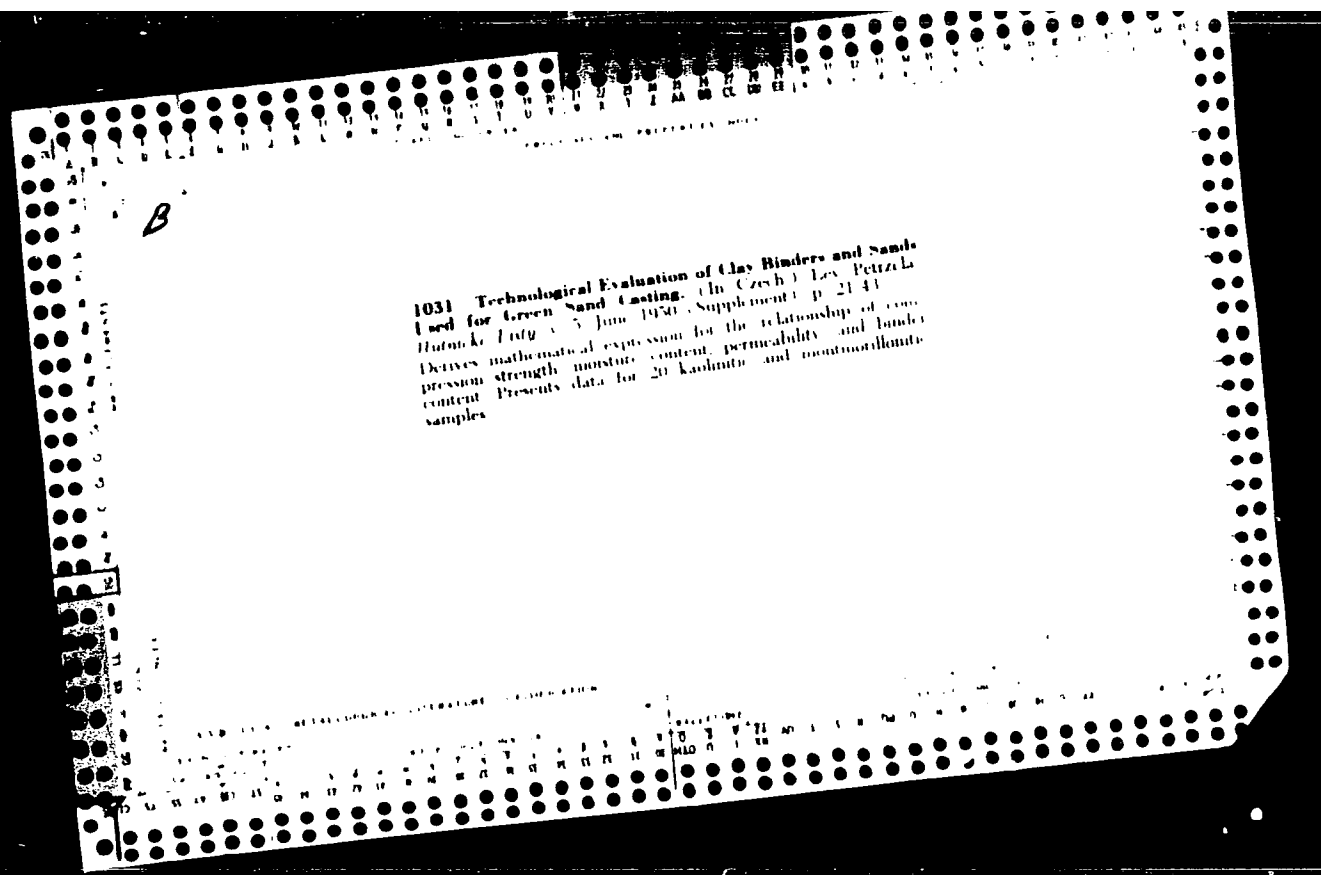
PEIRZELA, L.,
ZELEZARNY, VITKOVICKY, (B.P., 654-817, 10.12.48. Czechoslov.,
12.12.47 and 5.2.48)











PFTRZELA, L.

Survey of the use of the chemical hardening method in foundries. p. 227.

SLAVARENSTVI. (Ministerstvo težke strojirensivi a Ceskoslovenske vedec
technicke spolocnosti pro kunicivni a slavarensivi) Praha, Czechoslovakia.
Vol. 7, no. 7, June, 1977.

Monthly List of East European Accidents, (JAI), LC, Vol. 1, No. 12, Dec. 1977.
Uncl.

PETRZELA, L.

Properties of molding materials in elevated temperatures. Tr. from the Czech.
p. 65. (Przeglad Odlewnictwa, Vol. 7, No. 3, Mar 1957, Krakow, Poland)

SC: Monthly List of East European Accessions (MEAL) LC, Vol. 6, No. 8, Aug 1957. Incl.

13499*
L. Petrzela
Hutnické listy, 1952, 11, p. 575-585.

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13499* Chemical Hardening Molding Mixtures. (Czechoslovakian) L. Petrzela, Hutnické Listy, v. 7, no. 10, Oct. 1952, p. 514-518, no. 11, Nov. 1952, p. 575-585. Describes sand additives which harden quickly under effect of CO₂.

[illegible]

PETRZELA, Lev

2

11530* (Burning-On of Molding Materials on Castings.)
Príspevek k štúdiu spečení. Lev Petrzel. Strojárstvo,
v. 2; Práce Československého Vědeckého Sborníku, v. 1,
no. 1, Feb. 1954, p. 1-12.
Method of testing; effect of C additions to sand and mold and
core coatings; evaluation of results by comparison. Diagrams,
micrographs, tables. 48 ref.

Petrzela, L.

1ST AND 2ND ORDERS	PROCESSES AND PROPERTIES INDEX	3RD AND 4TH ORDERS
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CONTRIBUTION TO THE PROBLEM OF PREPARATION OF MOULDING SANDS. L. Petrzela. (Hunticks Listy, 1950, vol 5, Sept., pp 353-359; Oct., pp. 406-412). In Czech. The author reviews the deficiencies of natural sands and the objects of sand preparation and reports the results of tests on processes of preparing sand. The influence of drying on the properties of green glauconite sand was determined after drying for 4 hr. at the temperatures 90°, 110°, 300° and 400° C. samples which had been prepared for periods of 5, 10, and 20 min. in a Simpson mixer. Compression strength and permeability decreased with increasing drying temperatures, but the decrease was reduced by a longer mixing time; the moisture content corresponding to optimum compression strength remained practically unchanged. The compression strength of specimens mixed in a rib type mixer, changed gradually from 900g/sq. cm. to up to 1250 g/sq. cm. after 1 hr. mixing; this reduced the optimum moisture content from about 6% to 4% after a mixing time of 60 min. In the case of sands mixed with bentonite maximum compression strength was obtained after mixing for 10 to 20 min.---Ead.

ADDITIONAL METALLURGICAL LITERATURE CLASSIFICATION		DETERMINATION
EDITION	DATE	EDITION
EDITION	DATE	EDITION

PETRZELA, LEV

✓ 15090* Exothermic Sleeves for Risers. Exothermické obkl-
ady naliká. (Czech.) Lev Petrzela; Stávkrenstol, v. 3, no. 8;
Práce Československého výzkumního ústavu stávkrenského, v. 2, no. 21,
Aug. 1953, p. 145-148.
Because of their control, best results are obtained with mix-
tures with aluminothermic reactions. Photographs, graphs, table.
23 ref.

Petrzela, Lev

CZECH

✓ The application of signs in foundation. Lev Petrzela.
(Větrnost lety vnitřní křeh. Brno, Československo, 1944).
Zpr. 8, 544-55 (1944).—A lecture. Jan Miska

PETRZELA, L. J.

Journal of Applied Chemistry
Vol. 4 Feb. 1954
Industrial Inorganic Chemistry

①
✓ Bentonite moulding materials in Czechoslovakia. L. J. Petržela
(Prace. Odlewn., 1953, 8, No. 4, 109—115; J. Iron Steel Inst., 1953,
175, 222).—The properties of bentonite moulding materials and
their influence on foundry output are discussed. The prep. of
such materials for Fe and steel castings is described. R. B. CLARKE

CONTRIBUTION ON THE PROPERTIES OF BENTONITE OF BRANANY (CZECHOSLOVAKIA).
 L. Petršela. (Hutnické Listy, 1948, vol. 3, No. 1, pp. 335-339). (In Czech).
 The properties of synthetic sands produced with bentonite from Branany are
 described and compared with those of the French bentonite Claesol FE. The
 author also studied the relation between the swelling capacity of bentonite
 and its mechanical properties and concludes that there is no direct relation
 between the two; conclusions on the mechanical properties of moulding sand
 should not be drawn from results of swelling tests alone. E.O.

p - 19

ASB-354 METALLURGICAL LITERATURE CLASSIFICATION

PETRZELA, LEV.

CZECH

5367* Dry Molding Mixtures With Quartz Sand. Formovani
anot na sušení s křemennými písky. (Czech.) Lev Petřela.
Strojnický, v. 2: Práce Československého Vědeckého Školení
ského, v. 1, no. 13, Dec. 1954, p. 89-100.
Evaluation of sand and clay mixtures for various casting condi-
tions for steel and gray iron. Graphs, tables. 83 ref.

PETRZEL, L.

Properties of Binding Materials at Elevated Temperatures. J. Petrzel. (Silicates, 1958, vol. 4, Oct., 81-70). Natural and synthetic mixtures were used. Compressive strength at high temp. with slow heating is highest with clay binders at 900-1000° C. Below 800° C. the strength is up to twice that after drying. Over 1000° strength decreases suddenly. Thermal expansion was found to depend little on binder. Quartz sand mixtures have expansions 1.5-2%. Arcclay mixtures 0.7%. Maximum strength for chemically hardened mixtures occurs between 450-550° C, over 800° it decreases. Conversion of calcium-bentonite to Na increases strength up to 3 times. The properties determined are related to crack and spall formation and to scale and sand inclusions. Behavior of chemically hardened mixtures is reviewed in relation to compressive strength. Spall formation is considered, and compressive strength measurements are recommended for control of natural and synthetic mixtures for dry sand and green sand castings. (53 references).

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RD 048

PETRZELKA, K.

New system of wages in the clothing industry. p. 18.

(Textil. Vol. 12, no. 1, Jan. 1957. Praha, Czechoslovakia)

SO: Monthly List of East European Accessions (EEAL) 10, Vol. 6, no. 10, October 1957. Incl.

PETRZELKA, K.

Intrafactory transportation in the textile industry. p. 254.

TEXTIL . (Ministerstvo lehkého průmyslu) Praha, Czechoslovakia,
Vol. 14, no. 7, July 1959.

Monthly List of East European Accession (EEAI), LC Vol. 9, no. 2,
Feb. 1960.

Uncl.

1000 1000, 1000.

specialization of workshops in the field of industry. . . .
III. Administrative leakers (russ.) (russ.) . . . 7.
july 1964.

1000: East European Allegations list, Vol. 6, No. 1, 1964.

PETRZHAK, G.T.

Oxalate compounds of quadrivalent uranium. A. A. Grinberg and G. T. Petrzhak. *Trudy Radiofiz. Inst. im. P. G. Khlopin, Khim. i. Geokhimp.* 7, 17-40 (1958).—High resistance of quadrivalent uranium oxalate to air oxidation was the reason of many investigations. The hydrated K salt $K_2[U(C_2O_4)_4] \cdot 6H_2O$ was prepd. The Ba salt was prepd. by $K_2[U(C_2O_4)_4] + 2BaCl_2 \rightarrow Ba_2[U(C_2O_4)_4] + 4KCl$ to give the heptahydrate that after some time lost a mol. of water to form the hexahydrate. The octahydrate of the Pb salt and the heptahydrate of the Cd salt were obtained in an analogous manner. Ca^{++} reacts with $K_2[U(C_2O_4)_4]$ to yield ppt. of $U(C_2O_4)_3$ and CaC_2O_4 . Magnetic susceptibility studies of $U(C_2O_4)_3$ and K, Ba, and Ca salts of $H_2[U(C_2O_4)_4]$ indicate 2 unpaired electrons. The absorption spectrum of $K_2[U(C_2O_4)_4] \cdot 6H_2O$ has max. 490.0, 560.0, and 895.8 m μ , and the complex salt shows predominate character of ionic bonding. G. Melnyk

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PETRZHAK, G.I.

GRINBERG, A.A.; PETRZHAK, G.I.

Investigation of the physical and chemical properties of tetravalent
uranium and thorium oxalates. Trudy Radiev.inst.AN SSSR 7:50-73 '56.
(Uranyl oxalates) (Thorium oxalate)

GRINBERG, A.,

Paris, 1 -24. Jan. 1957.

PETRZHAK, G. I.

27 7
Slightly soluble compounds of quadrivalent uranium obtained with the aid of rongalite. A. A. Grinberg, L. R. Nikol'skaya, G. I. Petzhak, B. V. Pilyayn, and R. M. Pilyayn (Radium Inst., Leningrad). *Zhur. Anal. Khim.* 12, 62-4 (1957). Sexivalent U taken as $UO_2(NO_3)_2$ was reduced to the quadrivalent form by rongalite in acid soln. The reduction was carried out in the presence of $H_2C_2O_4$ so that the oxalate pptd. The reaction was carried out in HNO_3 of final concn. 1N, in the presence of $H_2C_2O_4$ 10% excess of theoretical value using for each mol. U 1 mol. rongalite plus 7-10% excess. With this procedure the soln. contained no more U than detd. by the soly. of the oxalate. Under similar conditions U was also pptd. as UF_4 .
M. Hersh

GRINBERG, A.A.; PETRZHAK, G.I.; NIKOL'SKAYA, L.Ye.; PTITSYN, B.V.; FILINOV,
F.M. [deceased]

New means of preparing tetravalent uranium derivatives. Trudy
Radiofiz. inst. AN SSSR. 8:166-169 '58. (MIRA 12:2)
(Uranium compounds)

V1 7 8 2 4 4 A 6 T

AUTHORS: Grinberg, A. A., Petrzhak, G. I., Levleya, I. I.

TITLE: **Studies in the Chemistry of Uranium- and Thorium-Oxalates**
(Issledovaniya po khimii oksalatov urana i toriya).

PERIODICAL: Zhurnal Neorganicheskoy Khimii, 1958, Vol. 3, No. 1, pp. 204-211 (USSR).

ABSTRACT: After a brief survey of literature the authors find that according to reference 6, the formulae of uranium oxalate (IV) and its derivatives should be doubled. Since these conceptions were not in mutual accordance with the theory of coordination and since they were not based on any physico-chemical data, the authors charged themselves with supplying this necessity.

Complex uranium oxalates. A salt $(UC_2O_4)_2 \cdot 2K_2C_2O_4 \cdot 5H_2O$ was obtained.

Barium salt, as well as new representatives of this series of compounds: lead-, cadmium-, and calcium-salts were isolated by double-exchange-reactions. The two latter representatives of the uranium-IV-derivates are "abnormal" red-violet colored. The molecular conductivity of the water solutions of the potassium salts was investigated for proving the coordination-structure of the oxalates of uranium.

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Studies in the Chemistry of Uranium- and Thorium-Oxalates.

74-127413

salts. Respective measurements prove that the potassium salt is a 5-ion-electrolyte which decays according to the scheme:

$K_4/U(C_2O_4)_4 \rightleftharpoons 4K^+ + /U(C_2O_4)_4^{4-}$. Previous test results - which were precisely determined meanwhile - together with data from literature for other 5-ion-electrolytes are shown in table 1. Since the pH of the potassium salt solutions hardly deviates from the pH of the water (figure 1) at the beginning, the authors, taking account of the conductivity concluded that the coordination number of uranium in potassium salt is equal to 8. Probable equations of the interaction of the $[U(C_2O_4)_4]^{4-}$ -ion with water are set up and the

further dissociation of the aquo-ions is explained. The pH consequently decreases in the course of time. The constants of stability of the afore-mentioned ions are evaluated (according to reference 12) and a value of the order 10^{-4} is obtained by potentiometric titration with HCl. The known salt $U_2(C_2O_4)_2 \cdot K_2C_2O_4 \cdot 8H_2O$ precipitates with this titration (after adding 1 mol HCl). The production of mixed salts was achieved with cerium and lanthanum (according to reference 13). There are metastable phases which gradually decay in the solution

Card 2/4

of uranium (IV) are not included in the formation of the 8-covalent bonds.

Comparison of the properties of $U(C_2O_4)_2 \cdot 6H_2O$ and $Th(C_2O_4)_2 \cdot 6H_2O$.

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001240630006-9"

The authors elaborated 2 new methods of determination of tetravalent uranium: 1) based upon an electrochemical method of $UO_2(C_2O_4)_2$ in presence of a surplus of oxalic acid. 2) by the action

Card 3/4

Studies in the Chemistry of Uranium- and Thorium-Oxalates.

of ronalite on uranyl salts in an acidic solution, a necessary surplus of oxalic acid. The molecular conductivity and the pH of the $U(C_2O_4)_3 \cdot 6H_2O$ resulting from the reaction, the water solubility (comparing for $Th(C_2O_4)_3 \cdot 6H_2O$) and the conductivity of aqueous solutions of various acids were investigated and a potentiometric titration with permanganate in a H_2SO_4 -medium was carried out.

2 potential differences occur with the titration (figure 2): a) with the oxidation of U^{IV} in U^{VI} , b) at the end of the oxidation of the oxalate-groups. It was found that $U(C_2O_4)_3 \cdot 6H_2O$ and $Th(C_2O_4)_3 \cdot 6H_2O$ behave quite different in water solutions. The much greater

of the acidity of uranium salt, compared with thorium salt, was confirmed by the action of gaseous pyridine on the two hexahydrates. The acid properties of uranium salt are finally confirmed by a lower solubility in diluted acids than in water. The difference in the properties of acidity between the derivatives of $U(IV)$ and $Th(IV)$ is explained by the relation of the ronalite.

There are 4 figures, 2 tables, and 12 references, 11 from the literature.

April 29, 1957.

Library of Congress.

SUBMITTED.
AVAILABLE.
Card 4/4

GRINBERG, A.A.; ~~PETRZHAK~~, G.I.; YEVTEYEV, L.I.

Preparation and properties of uranium-oxalic acid salts of rare
earth elements. Radiokhimiya 1 no.3:300-308 '59.
(MIRA 12:10)

(Oxalic acid) (Oxalatouranates)

YEVTEYEV, L.I.; PETRZHAK, G.I.

Synthesis of the sulfate of quadrivalent uranium using ronzalite.
Radiokhimiia 1 no.5:581-582 '59. (MIRA 13:2)
(Sodium formaldehydesulfoxylate)

GRINBERG, A. A. ; PETRZHAK, G. I. ; YEVTEYEV, L. I.

Instability constants of oxalate complexes of uranium. Radiokhimiya
2 no.4:505-506 '60. (MIRA 13:9)
(Potassium uranium oxalate)

24086

3/186/60/002/006/001/02-

A051/A123

21 4200

AUTHORS Grigor'ev, A.A., Vykhovskiy, P. N., Nikol'skiy, L. Ye.;
Buzdakov, S.I.

TITLE The formation of tetra-valent uranium oxalate from solutions
using rongalite

PERIODICAL Radiokhimiya, v. 2, no. 6, 1960, 687 - 690

TEXT The authors have established the conditions of precipitation of
tetra-valent uranium in dilute solutions using rongalite. It is shown that in
the formation of low quantities of uranium the rongalite concentration in the
solution must be below 0.1 M. Experiments were conducted to establish
the uranium losses in the filtrates after the formation of tetra-valent uranium
oxalate. The results obtained show that at the given concentrations about
4 mg of uranium are left in the solution, whereby the losses seem to be con-
siderably reduced. The experiments carried out for establishing the conditions of low-
quantity formations of uranium showed that at low concentrations of uranium in

Card 1/3

24086

The formation of tetra-uranium...

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A051/A129

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The solution for the reaction must be increased as compared to the theoretically computed quantity. The deviation of the stoichiometry under these conditions is due to the low rate of uranium reduction and the strong effect of the side reactions taking place at the same time. Further experiments were conducted to determine the uranium losses resulting from the addition of small quantities of bongalite to the uranium solution already reduced with metallic zinc. The quantity of uranium was determined with an accuracy of 0.1 mg by dissolving the residue in sulfuric acid and titrating it with potassium permanganate. Normal results were obtained for the reaction at a ratio of



1:2. The quantity of oxidized uranium was determined from the difference, correcting for solubility. The oxidation of uranium in the third series of experiments is thought to take place as a result of certain intermediate compounds formed in the interaction of bongalite and air oxygen. When large quantities

Card 2/3

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A051/A129

The formation of tetra-valent

of rongalite are present, this phenomenon does not occur. There are 3 tables and 4 references: 3 are Soviet-bloc and 1 non-Soviet-bloc. The reference to the English language publication reads as follows: W. H. Reas. The Transuranium Elements, 423, N.Y., 1949.

SUBMITTED: October 26, 1959.

Card 3/3

5/166/62/004/003/006/022
E071/E433

AUTHORS: Grinberg, V.V., Petrzhak, G.I., Lozhkina, G.S.

TITLE: Some new salts of uranyloxalic acid

JOURNAL: Radiokhimiya, v.4, no.3, 1962, 289-295

TEXT: The authors investigated the synthesis of salts of uranyloxalic acid with complex cations. Initially it was attempted to obtain compounds with cations $[\text{Co}(\text{NH}_3)_6]^{3+}$ and $[\text{Co}(\text{NH}_3)_4(\text{H}_2\text{O})_2]^{3+}$ and anion $[\text{U}(\text{C}_2\text{O}_4)_4]^{4-}$ by the double exchange reaction in a neutral medium. However, these compounds were not isolated, due to their rapid decomposition caused by the oxidizing-reducing interaction between the components of the complex formed. Subsequently the reaction between chromium-hexacarbamide chloride (III) and potassium uranyloxalate was tried. A new complex compound chromium hexacarbamide uraneate $[\text{Cr}(\text{N}_2\text{H}_4\text{CO})_6]_4[\text{U}(\text{C}_2\text{O}_4)_4]_3 \cdot 11\text{H}_2\text{O}$, in which the anion of uranyloxalic acid is bound to trivalent complex chromium cation, was synthesized. It was shown that this compound is stable in air and can be obtained either with an excess of $\text{K}_4[\text{U}(\text{C}_2\text{O}_4)_6]$ or of $[\text{Cr}(\text{N}_2\text{H}_4\text{CO})_6]\text{Cl}_3$. The compound is little soluble in water (30 mg per litre) and

Card 1/2

CHENEPKO, A.A., PETRZHAK, G.I.; Prinimal uchastiye YEVTEYEV, L.I.

Additional data on the solubility of tetravalent uranium
oxalate. Radiokhimiya 5 no.3:319-329 '63. (MIRA 16:10)

(Uranyl oxalate) (Solubility)

GRINBERG, A. A.; PETRZHAK, G. I.; LOZHKINA, G. S.

New salts of oxalotouranic acid. Radiokhimiia 4 no.3:289-295
'62. (MIRA 15:10)

(Oxalotouranates)

ADAMOV, V.; GRAUDYN', L.[Graudina, L.]; PETRZHAK, K.; SOROKINA, A.

Gamma rays from inelastic scattering of 2.95 Mev. neutrons in La^{139} .
Vestis Latv sk no.5:61-64 '61.

The determination of the decay constant of thorium by the Wynn Williams counter $K = 1.4 \times 10^{-10}$ per year and radium $T = 8.8 \times 10^4$ (1000000) 1500 of the const. by the recording galvanometer and by the phytatron counter gives comparable values of 5.17×10^{-10} years $\pm 4\%$. The half-life period is 1.34×10^{10} years $\pm 4\%$. The effect of emanation is eliminated in these expts. The accuracy of the measurement depends on the accuracy of weighing the samples. H. M. L.

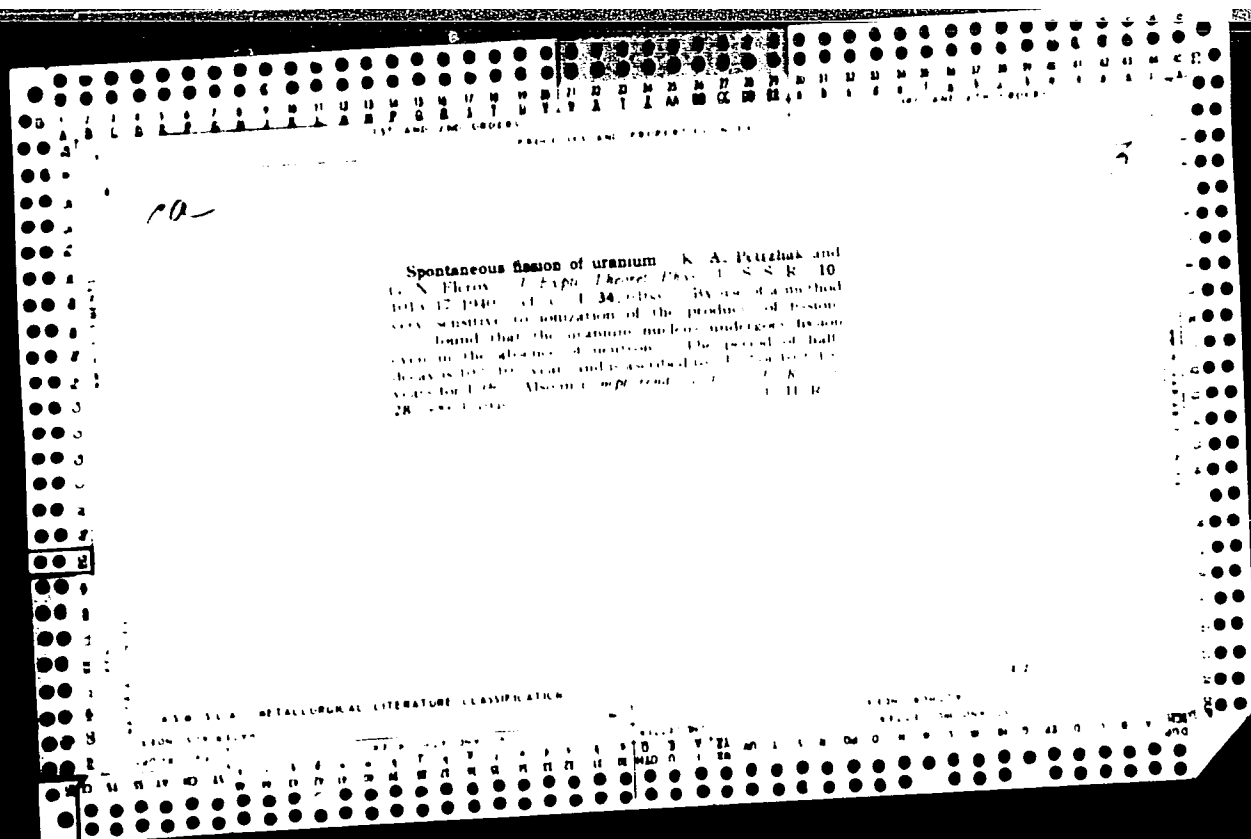
Possibility of the formation of a radioactive nuclei in
 the decay of radioactive chlorine. S. N. Dmitriyev, K.
 A. Petrizhuk and N. A. Perinoy. *J. Exptl. Theoret.*
Phys. (U. S. S. R.) 9, 1654 (1959). The disintegration
 of radioactive chlorine obtained by neutron bombardment of
 NaCl and CHCl_3 was studied in a Wynn-Williams and a
 low-pressure Wilson chamber, resp. The α -radioactivity
 found by Tolby, Peterson and Latimer of C. A. 29, 7179,
 could not be found. Conclusion: The α primarily formed
 is stable and the possibility of the simultaneous existence
 of a stable and an α -radioactive chlorine isotope is
 not confirmed.

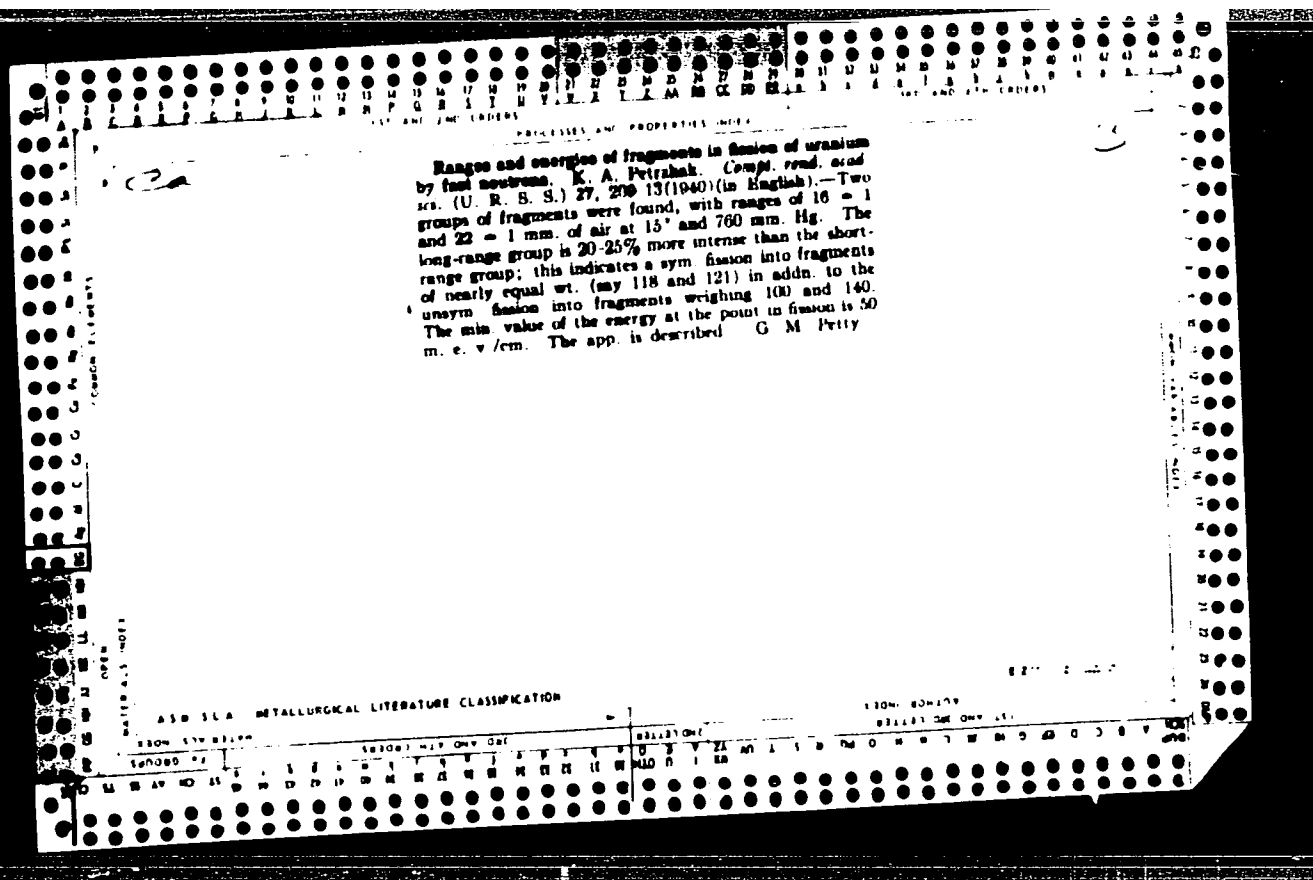
3

Spontaneous fission of uranium. V. S. Etkin and
A. A. Petrichuk. Zh. fiz. i matem. fiz. 1969, 13, 100.
See also 34, 1000.

Institute for Pure and Applied Physics, Leningrad

ASB 55A METALLURGICAL LITERATURE CLASSIFICATION





PETRAZHAK, K. A.
USSR/Physics - Radiation Rate

FD-2827

Card 1/1 Pub. 153-10/30

Author : Petrazhak, K. A. and Bak, M. A.

Title : Determination of Radiation Rate Falling on a Round Target from a
Round Source

Periodical : Zhur Tekh. Fiz, 25, 636-643, 1955

Abstract : The problem is solved in the form of two series which should be
applied according to ratios of target and source radii. The ir-
radiation rate on the target is tabulated in relation to the dis-
tance from source for nine ratios of target and source. Three US
references.

Institution :

Submitted : June 23, 1955

PETRZHAK, K. A.

Isotopic composition of uranium in meteorites. K. A. Petrzhak, I. N. Semenyushkin, and M. A. Bar. *Geokhimiya*, 1956, No. 2, 27-36. Report of a study the purpose of which was to det. accurately the isotopic U compn. of meteorites and to compare it with that for the earth. The U^{235}/U^{238} ratio was used as a criterion of the isotopic compn. of U. The method used was based on the different probabilities of fission of U^{235} and U^{238} under the action of thermal and fast neutrons. Cross-sectional diagrams are shown for the double ionization chamber used and for the part of the nuclear reactor in which the samples were placed. A diagram showing arrangement of the exper. app. employed is provided. The isotopic compn. of U was detd. in 4 iron meteorites and in 4 stone meteorites. For the stone meteorites the isotopic compn. of U in the chondrules and cementing material was detd. Samples of earth origin were prepd. from pitchblende. Detns. had a statistical accuracy of $\pm 2-4\%$. All values of K were found to be equal to 1.00 ± 0.02 , where $K = K_1/K_2$, and where K_1 is the intensity of fission of U by thermal neutrons and K_2 is that produced by fast neutrons. The isotopic compn. of U in meteorites and for the earth was found to be the same within an accuracy of $\pm 2-4\%$. It was concluded that (1) the U on earth and in meteorites apparently has a common origin, (2) the law of radioactive decompn. of U on earth and in meteorites is the same, and (3) in places of occurrence of meteorites, neutron fields are absent. 23 references. Gladys S. Mary

Petrzhak, K.A.

334.23 : 538.11 : 538.33 : 538.58
 5455. RADIATION FROM A SPHERE IN THE PRESENCE
 OF SELF ABSORPTION. M.A.Bak, K.A.Petrzhak and
 Yu.F.Romanov.
 Zh. tekhn. Fiz., Vol. 25, No. 2, 379-84 (1950). In Russian.
 A mathematical treatment of a sphere with a uniform
 rate of radiation throughout and a constant coefficient of
 absorption (μ). It is shown that the effect of absorption de-
 pends on μR where R is the radius and a table is given for
 $0 \leq \mu R \leq 10$. J.M.Hough

3

Petrzhak, K. A.

V4932

DETERMINATION OF ABSOLUTE YIELD FROM NEUTRON

SOURCES. M. A. Bak, K. A. Petrzhak, and Yu. F. Romanov.

Uspekhi Fiz. Nauk 55, 647-649 (1955) Apr. (in Russian)

A review is given of works published up to 1955 on the question of determining the absolute number of neutrons emitted per sec by neutron sources. Results of the works are completely tabulated giving the comparative data on the type of sources, quantity of Ra or Ra, and the Be absorbers, the type of detectors, and the neutron yield from the Ra + Be and Ra + Be sources. (N.V.J.)

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PETRZHAK, K.A.

Spontaneous fission of heavy nuclei. Atom.energ.supplement no.1:152-181
'57. (MIRA 10:10)

(Nuclear fission)

PETRZHAK, K.A.; SEMENYUSHKIN, I.N.; BAK, M.A.

Uranium isotopes in meteorites. Biul.Kom. po opr.abs.vozr.geol.
form. no.2:38-40 '57. (MIRA 10:4)

1. Radiyevyy institut im. V.G. Khlopina AN SSSR.
(Uranium--Isotopes) (Meteorites)

12C-4-3/35

AUTHORS: Bugorkov, S.S., Melnik, I.Z., Petrov, K.A., Yakovlev, V.A.
and Yakunin, M.I.

TITLE: Ionisation Chambers for Alpha Particle Counting
(Ionizatsionnye kamery dlya scheta al'fa-chastits)

PERIODICAL: Priroda i Tekhnika Eksperimenta, 1957, No.4,
pp. 16 - 19 (USSR)

ABSTRACT: The construction and properties of 5 ionisation chambers
for alpha particle counting are described.

no.1: A universal camera for alpha particles emitted within
a solid angle of 2π (Fig.1). This camera is used for
measurements on alpha-active materials deposited on one or
both sides of a thin plate. It can also be used to estimate
the degree of alpha-activation of the inner surfaces of
hemispherical platinum cups after various chemical procedures.

no.2: A camera for measurements in a solid angle which is
less than, or equal to, 2π (Fig.4).

No.3: A camera for measuring alpha activities of liquids
(Fig.5).

No.4: A camera with a solid angle $(0.01 - 0.001) \times 2\pi$ (Fig.6).

No.5: An argon filled camera (Fig.7). This is used to
measure intensities of the order of 2×10^5 counts/min and also

Card1/2

Petrzhak, K.A.

USSR/Physical Chemistry: Some Questions Concerning Systemic Structure of Matter.

Abstr Jour: Referat. Zhurnal Khimiya, No 2, 1958, 3480.

Author : N.P. Avdonin, K.A. Petrzhak.

Inst : Leningrad Institute of Technology, Leningrad.

Title : Study of (, n) and (, f) Reactions of Th^{232} by Radiochemical Method with -Quantum Energy up to 13 Mev.

Orig Pub: Tr. Leningr. tekhnol in-ta im. Lensovet, 1957, vyp. 40, 5-15.

Abstract: Photoreactions of Th^{232} were studied. A betatron was used as the source of γ -radiation. The Th specimens were irradiated with γ -quanta of $7 \leq E \leq 13$ Mev. The number of fission events was measured by the activity of Mb^{99} and Ce^{143} with $T_{1/2} = 67$ and 33 hours respectively. The integral cross section ratio of two concurring processes - photofission and photonuclear emission - was measured. Radioactive chains with mass numbers 143, 117, 113, 99, 97 and 91 were separated.

Card : 1/1

-3-

PA - 2718

AUTHOR
TITLE

PETRZHAK, K.A., BAK, M.A., FERSMAN, B.A.,
Determination of the Absolute Number of Neutrons emitted by a Radium-
Beryllium Source by Comparison with a Photoneutron-Deuterium Source.
(Opredeleniye absolutnogo chisla neytronov, ispuskayemykh radiy-berili-
yevym istochnikom, sravneniye s fotoneytronnym deyteriyevym istochni-
kom., Russian)
Atomnaya Energiya, 1957, Vol 2, Nr 4, pp 319-326 (U.S.S.R.)
Received 5/1957
Reviewed 6/1957

PERIODICAL.

ABSTRACT

The authors develop a comparatively simple method realizable in any laboratory for the gauging of neutron sources. For the purpose of determining this absolute neutron number two problems have to be solved. 1) Determination of the absolute number of photoprotons produced in a given volume with gaseous deuterium at a known pressure and at a known temperature. The gas is then replaced by heavy water and by this a so-called primary neutron standard is produced which emits a certain number of neutrons. 2) By comparison with this primary standard the absolute number of neutrons is obtained which is emitted within the time unit of the Ra-Be-source. Experimental arrangement is discussed on the basis of a drawing. The ionization chamber consists of a nickel hollow sphere and a hollow brass finger serves as collecting electrode. Recording of the photoprotons is given step by step. The relative neutron intensities of the γ -D-source and of the Ra- α -Be source were determined by comparison of the integral spatial distributions of slow neutrons in water. The neutrons were slowed down by water which is con-

Card 1/2

PETRAZHHK K. I.

RUBIN. / Nuclear Physics - Installations and Instruments. Methods of Measurement and Research

Ann Jour : Vop. Zhur - Fizika, No. 4, 1957, 1-4 (1957)

Author : Bak M.A., Gorskikh G.V., Matviyenko V.I., Petrizik K.I.,
Rozhnov Yu.F.

Inst : Radium Institute, Academy of Sciences, USSR, Leningrad

Title : Radon Neutron Sources

Orig Pub : Bul. Inst. politechn. Insi, 1957, 3, No 1-2, 47-54

Abstract : By measuring the spatial distribution of the neutron density in H_2O , the authors have determined the power and the average energy of the neutron sources Ra-Be, Ra-B, Ra-C, Ra- CaF_2 , Ra- H_2O , Ra- Li , and Ra- Si (the α reaction). The absolute neutron yield from the various sources was determined by comparing the integral distributions of the slowed-down neutrons from the investigated and from a standard Ra-Be source. The mean energy of the neutron spectra was estimated from the magnitude of the relaxation length (L), determined from the measurements of the distribution of the density of the

Card : 1/2

"APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001240630006-9

1. The following information was obtained from a review of the

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2. The following information was obtained from a review of the

APPROVED FOR RELEASE: 06/15/2000

CIA-RDP86-00513R001240630006-9"

STARIK, I.Ye.; PETRZHAK, K.A.; SHATS, M.M.; SEMENYUSHKIN, I.N.; BAK, M.A.

Isotopic composition and abundance of uranium in meteorites.
Meteoritika no.16:126-130 '58. (MIRA 11:8)
(Meteorites) (Uranium)

AUTHORS: Kovrigin, B.S., Petrzhak, K.A. 307. 89-4 6-6/30

TITLE: Investigation of the Statistical Distribution of Events of Spontaneous Fission of U^{238} on the Basis of the Energies of Two Fission Fragments (Issledovaniye statisticheskogo raspredeleniya aktov spontannogo deleniya U^{238} po energiyam dvukh oskolkov)

PERIODICAL: Atomnaya energiya. 1958. Vol. 4. Nr 6. pp 547-554 (USSR)

ABSTRACT: By means of an apparatus consisting of a double ionization chamber, 2 amplifier channels, a coincidence scheme, and a double-beam momentum oscillograph the kinetic energy of each of the two fission fragments was measured namely for 780 cases of the spontaneous fission of U^{238} and about 4500 cases of the fission of U^{235} with slow neutrons.

On the basis of these data the statistical distribution of the acts of spontaneous fission and of fission with slow neutrons over the energy of the two fission fragments was drawn. With the aid of this distribution the following curves were constructed for both kinds of fission:

Card 1/4 1.) Energy spectrum of all fission fragments, both light and heavy.

Investigation of the Statistical Distribution of Events
of Spontaneous Fission of U^{238} on the Basis of the
Energies of Two Fission Fragments

Doc. 69-4-6 6/30

- 2.) Distribution of the axes of fission over the total kinetic energy of both nuclear fragments.
- 3.) Mass distribution of fission fragments.
- 4.) Dependence of the total kinetic energy of both fission fragments on the mass ratio of fission fragments.
- 5.) Dependence of the kinetic energy of light and heavy fragments on the total kinetic energy.

For both kinds of fission the respective curves take the same course.

The following are the characteristic properties of the fission spectra:

Card 2/1

Investigation of the Statistical Distribution of Events
of Spontaneous Fission of U^{238} on the Basis of the
Energies of Two Fission Fragments

SOV.89-4-6-6/30

	Spontaneous fission		Fission with slow neutrons	
	Light fragments	Heavy fragments	Light fragments	Heavy fragments
Most probable energy in MeV	90.0±0.3	54.4±0.4	92.3±0.12	56.6±0.15
Half width of peak in MeV	7.5	21.1	19.2	23.5
Ratio of most probable energies	1.6		1.63	
Ratio between the peak of light and that of heavy fragments	1.13		1.15	
Ratio between the height of the saddle separating the peaks and the height of the peak of light fission fragments	0.33		0.42	

Card 3/4

Investigation of the Statistical Distribution of Events of Spontaneous Fission of ^{238}U on the Basis of the Energies of Two Fission Fragments

DOI: 89-4-6-6/30

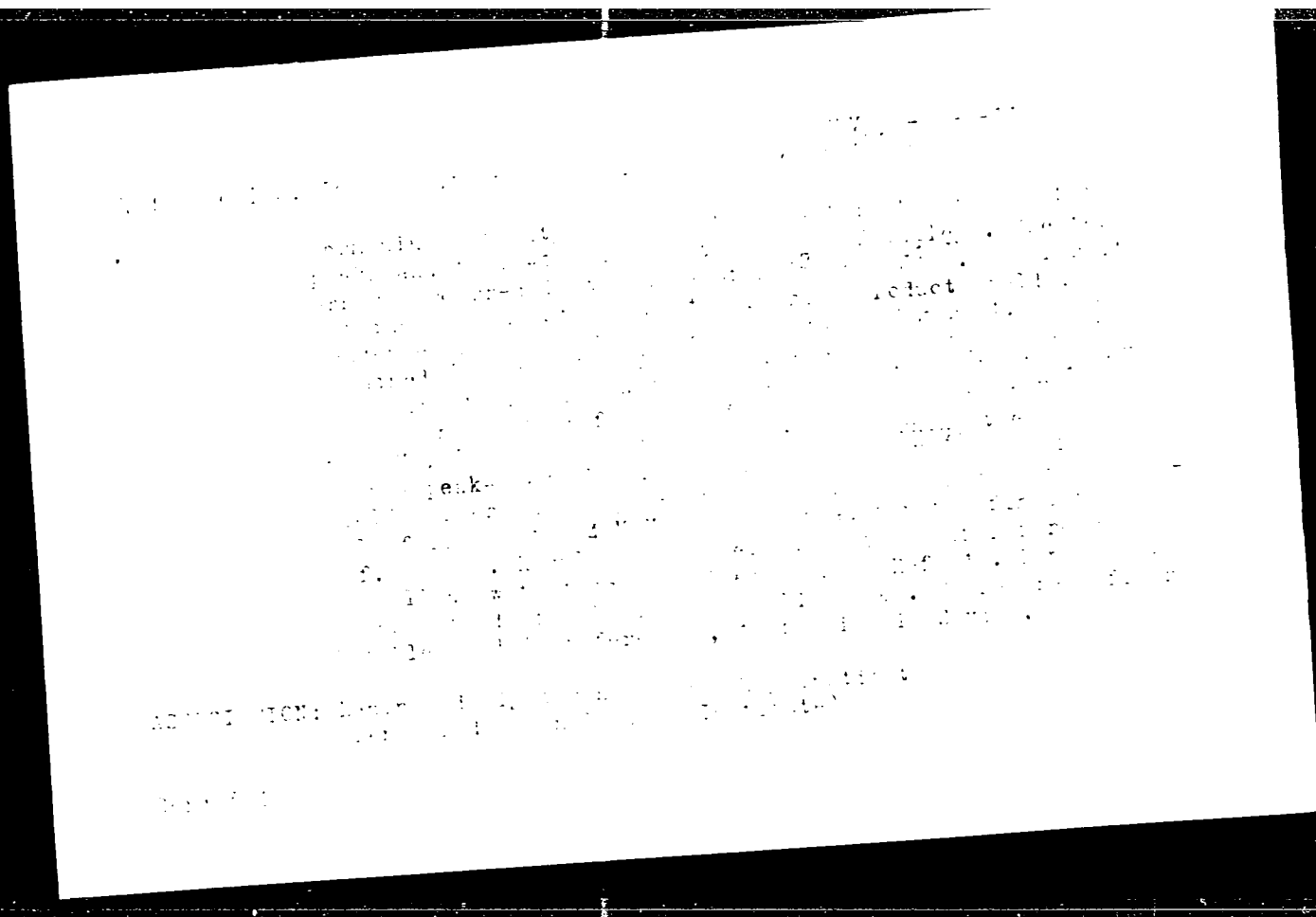
The author has 16 references

SUBMITTED

1. 7. 1989

1. Investigation of the statistical distribution of events of spontaneous fission of ^{238}U on the basis of the energies of two fission fragments (Radioactive substances and their properties)

Card 4/



DE + K ZHAB, + 1

BARON, N.M.; KVIAT, E.I.; PODGORNAYA, Ye.A.; PONOMAREVA, A.M.; RAVDEL',
A.A.; TIMOFEEVA, Z.N. Prinimal uchastiye VASIL'YEV, I.A..
MISHCHENKO, K.P., red.; PETRZHAK, K.A., red.; LOBINA, N.K., red.;
LEVIN, S.S., tekhn.red.; FOMKINA, T.A., tekhn.red.

[Short reference handbook of physicochemical constants] Kratkii
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Beklyady sovetskikh uchenykh: polucheniye i primeneniye izotopov (Reports of Soviet Scientists: Production and Application of Isotopes). Moscow, Atomizdat, 1960. 508 p. (Series: ITs; Trudy, vol. 6) 8,000 copies printed.

Eds. (Title page). G.V. Fordyushev, Academician, and I.I. Morikov, Corresponding Member, USSR Academy of Sciences; Ed. (Title book): I.D. Andreyenko, Tech. Ed.: I.D. Andreyenko.

PURPOSE: This book is intended for scientists, engineers, physicists, and biologists engaged in the production and application of isotopes. It is a practical treatise for professors and graduate students of higher technical schools where nuclear science is taught, and for the general public interested in atomic science and technology.

COVERAGE: This is volume 6 of a 6-volume set of reports delivered by Soviet scientists at the Second International Conference on the Peaceful Use of Atomic Energy held in Geneva from September 1 to 13, 1958. Volume 6 contains 52 reports on 1) modern methods for the production of artificial active isotopes and their application in various fields of science and technology; 2) the use of isotopes in the field of agriculture, as at work, at building, and agriculture; 3) the use of isotopes in the field of medicine; 4) the use of isotopes in the field of biology; 5) the use of isotopes in the field of physics; 6) the use of isotopes in the field of chemistry; 7) the use of isotopes in the field of geology; 8) the use of isotopes in the field of astronomy; 9) the use of isotopes in the field of meteorology; 10) the use of isotopes in the field of oceanography; 11) the use of isotopes in the field of geophysics; 12) the use of isotopes in the field of geodesy; 13) the use of isotopes in the field of cartography; 14) the use of isotopes in the field of navigation; 15) the use of isotopes in the field of communication; 16) the use of isotopes in the field of defense; 17) the use of isotopes in the field of industry; 18) the use of isotopes in the field of commerce; 19) the use of isotopes in the field of culture; 20) the use of isotopes in the field of education; 21) the use of isotopes in the field of sports; 22) the use of isotopes in the field of recreation; 23) the use of isotopes in the field of tourism; 24) the use of isotopes in the field of travel; 25) the use of isotopes in the field of transportation; 26) the use of isotopes in the field of communication; 27) the use of isotopes in the field of defense; 28) the use of isotopes in the field of industry; 29) the use of isotopes in the field of commerce; 30) the use of isotopes in the field of culture; 31) the use of isotopes in the field of education; 32) the use of isotopes in the field of sports; 33) the use of isotopes in the field of recreation; 34) the use of isotopes in the field of tourism; 35) the use of isotopes in the field of travel; 36) the use of isotopes in the field of transportation; 37) the use of isotopes in the field of communication; 38) the use of isotopes in the field of defense; 39) the use of isotopes in the field of industry; 40) the use of isotopes in the field of commerce; 41) the use of isotopes in the field of culture; 42) the use of isotopes in the field of education; 43) the use of isotopes in the field of sports; 44) the use of isotopes in the field of recreation; 45) the use of isotopes in the field of tourism; 46) the use of isotopes in the field of travel; 47) the use of isotopes in the field of transportation; 48) the use of isotopes in the field of communication; 49) the use of isotopes in the field of defense; 50) the use of isotopes in the field of industry; 51) the use of isotopes in the field of commerce; 52) the use of isotopes in the field of culture.

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